

Impurity types, concentration influence hydraulic design

Klaus D. Kaufmann
ILF Consulting Engineers
Munich



Real CO₂ streams—those from CO₂-capture plants likely to contain impurities as opposed to pure CO₂ streams—will likely contain at least 95 mole % CO₂ but will also contain impurities generated in the individual power plant and carbon capture-related facilities.

The first part of this article (OGJ, Apr. 12, 2010, pp. 39) described in detail methods for determining steady-state pressure and temperature profiles of such CO₂ streams. The conclusion, presented here, addresses the expected influence of impurities present in real CO₂ streams on the hydraulic pipeline layout and presents an overview

diagram enabling a first estimation of the most economic pipeline diameter, depending on intended CO₂ throughput rates.

Background

Type and concentration of the impurity components contained in the CO₂ stream will influence the hydraulic design of a pipeline system transporting real CO₂ streams, which depend on a series of considerations like:

- Power plant fuel type and carbon-capture technology.
- Health-related safety considerations referring to the maximum allowable concentration of toxic CO₂ stream components (e.g., H₂S, SO₂) in hypothetical leak situations.

CO₂ PIPELINES— Conclusion

of the pipeline steel, hydrogen-induced cracking, or sulfide stress cracking (which can be mitigated by appropriate pipe material selection).

- Storage requirements (e.g., concentration limitation of oxygen and noncondensable components).
- Limitation of the amount of economically usable additional components transported (e.g., thermal usage of hydrogen or methane).
- Limitation of the amount of additional components in order to minimize friction pressure losses or losses of pipeline transportation capacity.
- Limitation of the concentration of additional components in order to minimize the amount of energy required in the pipeline system's compression and transportation stations.

Impurity sources

The process or power plant application for combustion of the primary fossil fuels—coal, oil, gas, biomass, or a mixture of these—determines the CO₂ capture techniques, which for power plant applications are characterized commonly as precombustion, postcombustion, or oxy-fuel processes (Table 1).

The processes mentioned may generate components appearing at different combinations and concentrations in the CO₂ streams captured, H₂S and SO₂ resulting from the fuel's sulfur content. Table 2 gives an overview on the concentrations of the impurities expected in dried CO₂ streams.¹

While the stream compositions giv-

CO₂ CAPTURE PROCESSES FOR POWER GENERATION

Table 1

| Process | Description |
|------------------|--|
| Postcombustion | Separated from power plant flue gases by amine or other process. |
| Precombustion | Integrated gasification combined cycle generation of synthesis gas, gas shift reaction to H ₂ + CO ₂ , separation of CO ₂ and H ₂ , combustion of H ₂ in power plant. |
| Oxy-fuel systems | Combustion of fuel with almost pure oxygen, recycle flue gas consisting mainly of CO ₂ . |

EXPECTED IMPURITY CONCENTRATION IN DRIED CO₂ STREAMS

Table 2

| Component | Coal-fired plants | | | Gas-fired plants | | |
|-----------------------------------|-------------------|----------------|------------|------------------|---------------|------------|
| | Postcombustion | Precombustion | Oxy-fuel | Postcombustion | Precombustion | Oxy-fuel |
| SO ₂ | <0.01 | 0 | 0.5 | <0.01 | 0 | <0.01 |
| NO _x | <0.01 | 0 | 0.01 | <0.01 | 0 | <0.01 |
| H ₂ S | 0 | 0.01-0.6 | 0 | 0 | <0.01 | 0 |
| H ₂ | 0 | 0.8-2.0 | 0 | 0 | 1.0 | 0 |
| CO | 0 | 0.03-0.4 | 0 | 0 | 0.04 | 0 |
| CH ₄ | 0 | 0.01 | 0 | 0 | 2.0 | 0 |
| N ₂ /Ar/O ₂ | 0.01 | 0.03-0.6 | 3.7 | 0.01 | 1.3 | 4.1 |
| Total | 0.01 | 2.1-2.7 | 4.2 | 0.01 | 4.4 | 4.1 |

en in Table 2 reflect the aspects of the capture processes, Table 3 shows the DYNAMIS specification² taking safety and toxicity limits into account.

The DYNAMIS report² also states, however, that this recommendation covers a capture process applied to coproduction of electricity and hydrogen and, further, care must be used in applying this quality recommendation to other types of capture processes.

Impurity influence

Estimating the influence of impurities on the pressure and temperature profiles of a CO₂ pipeline system and on the power demand of the initial compression stations and potentially installed intermediate transportation station(s) requires estimating the influence of impurities on vapor pressure-critical pressure, density, viscosity, specific heat capacity, Joule-Thomson coefficient, and isentropic p-T-relationship.

The published data on the influence of impurities on CO₂ stream properties, the applicability of existing equations of state, and the applicable mixing rules and parameters data are, however, limited.^{3,4}

The Polytec report provides example estimates for pressure and temperature-dependent density, dynamic viscosity, and vapor pressure values.³ The REF-PROP program from National Institute of Standards and Technology obtained the data used by the report, referring to the statement by NIST that the program uses the most accurate equations of state currently available. The report³ comprises a compilation of available measurement data on pressure vs. temperature and vapor-liquid equilibrium data of mixtures of CO₂ with other components.

Table 4 presents the influence of

CO₂ STREAM SPECIFICATIONS, DYNAMIS

Table 3

| Component | Aquifer | Enhanced oil recovery | Remark, limitation ¹ |
|------------------|------------------------------------|-----------------------|------------------------------------|
| H ₂ O | 500 ppm | | Technical aspects ² |
| H ₂ S | 200 ppm | | Health, safety considerations |
| CO | 2,000 ppm | | |
| O ₂ | <4 vol % | 100-1,000 ppm | Technical aspects ³ |
| CH ₄ | <4 vol % | <2 vol % | Reference to ENCAP project |
| N ₂ | <4 vol %, all noncondensable gases | | |
| Ar | | | |
| H ₂ | | | Reduction recommended ⁴ |
| SO _x | 100 ppm | | Health, safety considerations |
| NO _x | 100 ppm | | |
| CO ₂ | >95.5% | | Balanced with other components |

¹Abridged remarks from DYNAMIS report. ²Expected in the future to range near 250 ppm. ³Range of EOR due to lack of practical experiments on O₂ effects underground. ⁴Due to energy content.

IMPURITY INFLUENCES AT 100 BAR, VARIABLE TEMPERATURE

Table 4

| Unit | CO ₂ , 100% | CO ₂ + 2% CH ₄ | CO ₂ + 2% H ₂ | CO ₂ + 2% N ₂ | CO ₂ + 2% Ar | CO ₂ + 2% SO ₂ | CO ₂ + 2% H ₂ S | CO ₂ + 2% O ₂ |
|--|------------------------|--------------------------------------|-------------------------------------|-------------------------------------|-------------------------|--------------------------------------|---------------------------------------|-------------------------------------|
| Relative density deviation compared with 100% CO₂ | | | | | | | | |
| 10° C. | % | 0.0 | -2.6 | -5.0 | -2.6 | -1.6 | 1.6 | -0.3 |
| 20° C. | % | 0.0 | -4.1 | -6.2 | -4.2 | -2.7 | 2.0 | -0.5 |
| 30° C. | % | 0.0 | -6.1 | -8.8 | -6.6 | -4.1 | 2.8 | 0.0 |
| Relative dynamic viscosity deviation compared with 100% CO₂ | | | | | | | | |
| 10° C. | % | 0.0 | -6.0 | -10.4 | -6.8 | -5.5 | — | 0.5 |
| 20° C. | % | 0.0 | -6.0 | -11.8 | -7.6 | -5.4 | — | 0.5 |
| 30° C. | % | 0.0 | -7.7 | -14.0 | -10.7 | -7.7 | — | 0.0 |
| CO₂ absolute vapor pressure, deviation compared with 100% CO₂ | | | | | | | | |
| 10° C. | bar | 45.0 | 5.3 | 17.6 | 9.8 | 8.3 | -1.3 | -0.4 |
| 20° C. | bar | 57.3 | 5.1 | 13.7 | 8.9 | 7.9 | -1.7 | -0.3 |
| 30° C. | bar | 72.1 | 6.8 | 8.5 | 6.3* | 4.3 | -2.7 | -0.4 |

*Extrapolated value.

SENSITIVITY CALCULATIONS VARYING INDIVIDUAL PROPERTIES ±10%

Table 5

| Property variation | Calculated pressure, temperature at pipeline end | | | |
|------------------------------|--|------------------|---------------|------------------|
| | Absolute values | | Differences | |
| | Pressure, bar | Temperature, °C. | Pressure, bar | Temperature, °C. |
| Base case | 91.41 | 27.09 | — | — |
| Density | | | | |
| 0.9 | 85.56 | 26.62 | -5.85 | -0.47 |
| 1.1 | 95.82 | 27.40 | 4.41 | 0.31 |
| Kinematic viscosity | | | | |
| 0.9 | 91.51 | 27.10 | 0.10 | 0.01 |
| 1.1 | 91.31 | 27.08 | -0.10 | -0.01 |
| Specific heat capacity | | | | |
| 0.9 | 91.74 | 26.44 | 0.33 | -0.65 |
| 1.1 | 91.13 | 27.66 | -0.28 | 0.57 |
| Joule-Thomson coefficient | | | | |
| 0.9 | 91.14 | 27.63 | -0.27 | 0.54 |
| 1.1 | 91.66 | 26.59 | 0.25 | -0.50 |
| Isentropic dp/dt coefficient | | | | |
| 0.9 | 91.34 | 27.20 | -0.07 | 0.11 |
| 1.1 | 91.48 | 26.98 | 0.07 | -0.11 |

impurities on density, viscosity, and vapor pressure of CO₂ streams at 100 bar with different temperatures, using an impurity concentration of 2%. These data were extracted graphically from the report's diagrams and are for illustration purpose only.

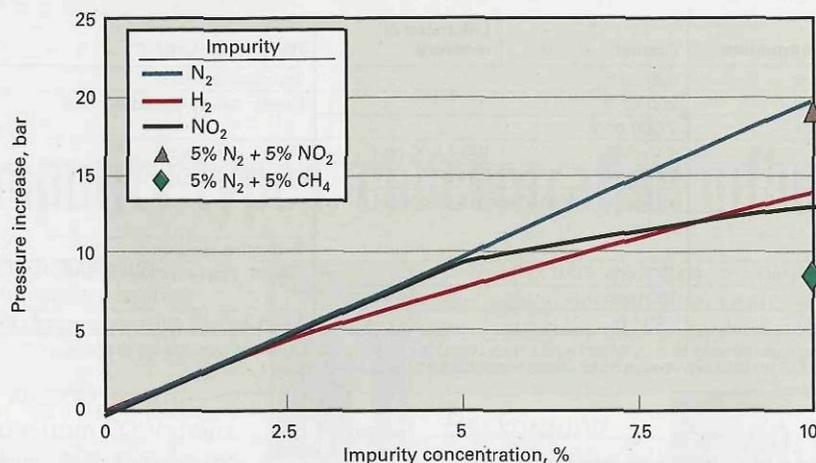
SO₂ is the only component increas-

ing stream density compared to pure CO₂, the estimated density for this mixture is very uncertain since no mixture parameters were available. H₂S has a minimal impact on the fluid density while H₂ has a large impact.

Impurities typically will reduce dynamic viscosity (Table 4).

CO₂ STREAM CRITICAL PRESSURE INCREASE, BY IMPURITY

Fig. 1



Impurities affect vapor pressure with the exception of H₂S and SO₂ (Table 4). The values for CO₂-SO₂ mixture are very uncertain, since mixing parameters were estimated and not based on actual measurement data. The presence of impurities also implies the presence of a two-phase region.³

Table 4 shows, for example, for a temperature of 30° C. (near CO₂'s critical temperature ~31° C.) the vapor pressure of a CO₂ mixture with 2% H₂ is about 8.5 bar higher than that of pure CO₂.

Literature addresses the influence of impurities on critical pressure.⁴ Fig. 1 presents the relationships and shows variations of critical pressure of CO₂ streams with different impurities.

Fig. 1 shows the increase of the critical pressure due to impurities is expected to remain moderate (<10 bar) if type and concentration of impurities remain in the ranges estimated in Table 2.

Fluid properties

Estimating the influence of impurities on the results of steady-state pressure and temperature profile calculations assumed modifications of relevant fluid properties of ±10%. Table 5 shows the results of related calculations

performed for the hypothetical CO₂ transportation system.

Table 5 shows variations of the CO₂ stream density due to impurities as representing a dominant factor in determining pressure losses along a pipeline system. Accurate determination of the CO₂ stream density regarding the presence of impurities therefore represents the major hurdle for reliable prediction of hydraulic pressure and temperature profiles along a new pipeline system for captured CO₂.

The development of a new CO₂

pipeline system requires estimation of the types and concentration ranges of impurity components of the CO₂ stream. Tables 1 and 2 estimates for this purpose depend on the technologies applied for power generation and carbon capture.

Table 4 and Fig. 1 can estimate the critical pressure of the transported CO₂ stream, defining the minimum operating pressure by considering the sufficient safety distance to the critical pressure.

Table 4 allows estimation of appropriate correction factors for density and viscosity of the CO₂ stream and after selection of an appropriate pipeline diameter, first hydraulic pressure and temperature profiles can be determined applying equations for consecutive pipeline sections from the pipeline system inlet to the system outlet presented in Part 1 of this article.

This procedure provides a straightforward methodology to develop basic hydraulic pipeline profiles for new CO₂ transportation systems, respecting also the influence of impurities on the calculated pressure and temperature profiles.

Economic aspects

After defining minimum operating pressure to avoid two-phase flow, minimizing specific CO₂ transportation costs, including initial investment cost and energy cost to compensate the friction losses, can estimate the optimum pipeline diameter.

Assuming a constant annual CO₂ throughput over the life of the project, the specific CO₂ transportation cost C_{sp} can be estimated with initial investment cost C_{inv}, the annuity factor a, the annual energy cost C_{en}, and the annual mass m_{yr} transported (Equation 1).

Annuity factor a is calculated as a function of interest rate i and number of operating years n (Equation 2).

Initial investment cost C_{inv} depends on parameters in-

EQUATIONS

$$C_{sp} = \frac{C_{inv} \cdot a + C_{en}}{m_{yr}} \quad (1)$$

$$a = \frac{i \cdot (1 + i)^n}{(1 + i)^n - 1} \quad (2)$$

MAIN INPUT DATA FOR RAW PIPELINE SYSTEM OPTIMIZATION

Table 6

| Specific pipeline transportation cost | Unit | Value |
|---------------------------------------|----------------|-------|
| Basic process data: | | |
| Density, average | kg/cu m | 770 |
| Kinematic viscosity, average | cst, sq mm/sec | 0.08 |
| Pump motor efficiency | % | 75 |
| Annual operating time | hr | 8,322 |
| Specific cost: | | |
| Specific pipeline system cost | €/lin.*m | 39.0 |
| Specific energy cost | €/Mw hr | 90.0 |
| Financial data: | | |
| Time period considered | years | 20 |
| Interest rate | %/year | 10 |

cluding pipe OD, design pressure, pipe WT, steel and coating delivery cost, and pipelaying cost. Estimates for a new CO₂ pipeline system in the 16-32 in. OD range with a design pressure of about 150 bar using typical western European costs of about €39/(inch*m) yield a price for a 24-in. OD pipeline of roughly 39*24 €/m = €936/linear pipeline m.

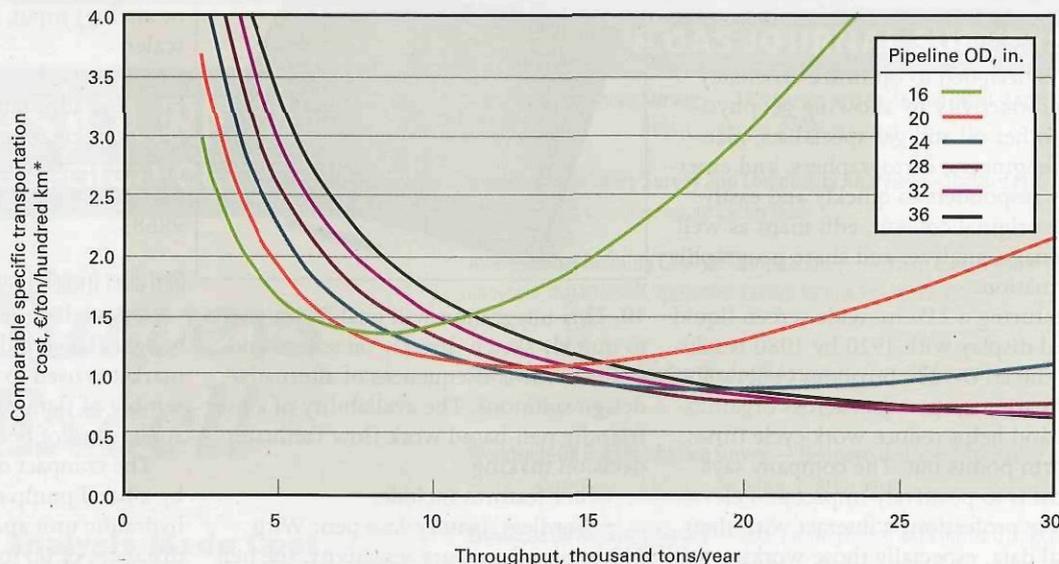
Annual energy costs C_{en} are based on a determination of diameter-dependent friction losses of the specific energy costs to operate the injection-transport stations and the annual operation time of the system.

Table 6 shows the main input data used for economic calculations, assuming the CO₂ stream is transported in dense phase at a density of 770 kg/cu m.

Fig. 2 shows the results of raw pipeline system optimization. For transportation of 10 million tons/year CO₂, a 20-in. OD pipeline system would represent the optimum techno-economic solution. The calculated specific transportation cost equals about €1.2/ton at 100 km transportation distance. A 24-in. OD pipeline system could, however, be even more suitable if a future CO₂ throughput expansion were intended (e.g., to 15 million tons/year).

The specific transportation cost shown in Fig. 2, however, reflects only the friction-loss related cost along the pipeline route. The specific cost to compress the CO₂ from the capture pressure level to the dense phase has to be added separately to the specific transportation cost.

DIAMETER OPTIMIZATION



*Initial compression in head station to 80 bar excluded. Inclusion adds about €9/ton energy cost, €2/ton annuity cost.

The specific shaft rated power demand for CO₂ compression assuming equal stage pressure ratios as well as isentropic and mechanical efficiencies of 0.80 and 0.90, respectively, is about 366 kJ/kg (1 bar/30° C. to 80 bar) and 21 kJ/kg (80 bar/40° C. to 130 bar). Estimates for the shaft rated power demand to compress 1,200 ton/hr CO₂ from 1 bar to 130 bar in the initial station measured about 122 + 7 = 129 Mw. Friction pressure losses inside the compressor station are not addressed.

Assumed specific shaft-rated energy cost of €90/Mw-hr yields a resulting specific compression energy cost of about €9.1/ton CO₂ (1-80 bar) and €0.53/ton CO₂ (80-130 bar). Specific annuity cost of the injection compression station is about €2/ton CO₂.

The intermediate transport station's shaft-rated power demand to increase pressure to 128 bar from 88 bar is about 2.4 Mw, about 1.9% of the compression power demand of the initial station.

The curves shown in Fig. 2 provide only a rough indication of optimum diameter for a given annual CO₂ throughput. Determining the optimum solution

in each individual case requires more detailed calculations. ♦

References

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2. "DYNAMIS Project No. 019672: Towards Hydrogen and Electricity Production with Carbon Capture and Storage," D 3.1.3, DYNAMIS CO₂ quality recommendations, June 21, 2007.
3. Polytec Report No. POL-O-2007-138-A, "State-of-the-Art Overview of CO₂ Pipeline Transport with relevance to Offshore Pipelines," Jan. 8, 2008.
4. Seevam, P.N., Race, J.M., and Downie, M.J., "Carbon dioxide pipelines for sequestration in the UK: an engineering gap analysis," Journal of Pipeline Engineering, Vol. 6, No. 3, pp. 140-141, September 2007.